- 2. A. J. STRAUSS and R. F. BREBRICK, J. Phys. Chem. Solids 31 (1970) 2293.
- 3. J. L. SCHMIT and C. J. SPEERSCHNEIDER, Infrared Phys. 8 (1968) 247.
- 4. H. RODOT, J. Phys. Chem. Solids 25 (1964) 85.
- 5. T. OKAZAKI and K. SHOGENJI, *ibid.* 36 (1975) 439.
- 6. B. E. BARTLETT, J. DEANS and P. C. ELLEN, J. Mater. Sci. 4 (1969) 266.
- D. K. HOHNKE, H. HOLLOWAY, E. M. LOGO-THETIS and R. C. CRAWLEY, J. Appl. Phys. 42 (1971) 2487.
- 8. S. A. IGNATOWICZ, Thin Solid Films 6 (1970) 299.
- 9. H. RODOT, Compt. Rend. 258 (1964) 6386.
- C. J. GILLHAM and R. A. FARRAR, Paper presented at International Colloquium on Semimetals and Narrow Gap Semiconductors, Nice (1973).
- 11. R. F. BREBRICK and A. J. STRAUSS, J. Phys. Chem. Solids 26 (1965) 989.

12. I. A. GRYADIL, N. I. DOVGOSHEI, A. P. ZHDANKIN and D. V. CHEPUR, Ukrain Fiz. Zhur. 16 (1971) 331.

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## High temperature thermal expansion of $ThO_2$ , MgO and $Y_2O_3$ by X-ray diffraction

There is an increasing need for knowledge of the physical properties of various oxides at high temperatures because of their importance as refractory materials. As limited data exist in the literature on the high temperature thermal expansion of these materials, the purpose of this note is to report some X-ray diffraction measurements of the thermal expansion of ThO<sub>2</sub>, MgO and Y<sub>2</sub>O<sub>3</sub>

between 500 and 1800° C.

The diffractometer, high temperature furnace chamber, and high temperature cell used in this work have been described previously [1]. The sample powder materials were purchased from Mitsuwa Chemical Company and Wakoh Chemical Company and were of nominal purity:  $ThO_2$ , 99.99%; MgO, 99.8%;  $Y_2O_3$ , 99.8%. All data were obtained using copper radiation, and a graphite monochromator in the diffracted beam was also used. The chamber was evacuated with rotary and

TABLE I Lattice parameters of ThO<sub>2</sub>, MgO and Y<sub>2</sub>O<sub>3</sub> at various temperatures

| Run<br>no. | ThO <sub>2</sub> |                          | MgO           |                          | Y <sub>2</sub> O <sub>2</sub> |                          |
|------------|------------------|--------------------------|---------------|--------------------------|-------------------------------|--------------------------|
|            | Temp.<br>(°C)    | Lattice<br>parameter (Å) | Temp.<br>(°C) | Lattice<br>parameter (A) | Temp.<br>( ° C)               | Lattice<br>parameter (A) |
| 1          | 25               | 5.592                    | 25            | 4.203                    | 25                            | 10.607                   |
| 2          | 614              | 5.621                    | 603           | 4.240                    | 619                           | 10.655                   |
| 3          | 806              | 5.636                    | 807           | 4.253                    | 814                           | 10.671                   |
| 4          | 1009             | 5.649                    | 996           | 4.262                    | 1007                          | 10.689                   |
| 5          | 1211             | 5.659                    | 1209          | 4.276                    | 1225                          | 10.710                   |
| 6          | 1407             | 5.670                    | 1416          | 4.287                    | 1408                          | 10.731                   |
| 7          | 1603             | 5.687                    | 1601          | 4.299                    | 1601                          | 10.757                   |
| 8          | 1782             | 5.698                    | 1739          | 4.308                    | 1764                          | 10.773                   |
| 9          | 25               | 5.601                    | 25            | 4.216                    | 25                            | 10.608                   |
| 10         | 508              | 5.619                    | 502           | 4.234                    | 543                           | 10.645                   |
| 11         | 714              | 5.633                    | 715           | 4.245                    | 704                           | 10.664                   |
| 12         | 905              | 5.641                    | 906           | 4.259                    | 917                           | 10.681                   |
| 13         | 1096             | 5.653                    | 1108          | 4.271                    | 1092                          | 10.697                   |
| 14         | 1323             | 5.664                    | 1317          | 4.281                    | 1303                          | 10.722                   |
| 15         | 1504             | 5.681                    | 1525          | 4.296                    | 1511                          | 10.743                   |
| 16         | 1711             | 5.693                    | 1704          | 4.304                    | 1685                          | 10.767                   |
| 17         | 1769             | 5.698                    | 1765          | 4.308                    | 1757                          | 10.772                   |

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oil diffusion pumps to a vacuum of  $3 \times 10^{-6}$  Torr. The lattice parameters were calculated from measured X-ray intensity patterns by employing the Cohen least-squares method [2]. The lattice parameters were estimated to be uncertain within about  $\pm 0.04\%$ .

The lattice parameter measurements on ThO<sub>2</sub>, MgO and  $Y_2O_3$  samples over the temperature range 500 to 1800° C are summarized in Table I. The lattice parameter at room temperature of each unheated oxide sample is almost identical to that reported in ASTM cards. The data in Table I are converted into % linear expansion and plotted in Fig. 1. They are best represented by the following equations;

ThO<sub>2</sub>: 
$$\Delta a/a_0 = 7.589 \times 10^{-6} (T - 298)$$
  
+ 1.552 × 10<sup>-9</sup> (T - 298)<sup>2</sup> (1)

MgO: 
$$\Delta a/a_0 = 1.241 \times 10^{-5} (T - 298) + 8.136 \times 10^{-10} (T - 298)^2$$
 (2)

$$Y_2 O_3 : \Delta a/a_0 = 6.537 \times 10^{-6} (T - 298) + 1.440 \times 10^{-9} (T - 298)^2,$$
(3)

where  $a_0$  is the lattice parameter (Å) at room temperature (25° C),  $\Delta a$  the change in lattice parameter between 298 K and temperature *T*, and *T* the absolute temperature. These equations are applicable over the temperature range observed, with a mean deviation of  $\pm 0.3 \times 10^{-3}$ . The results obtained in this work are in good agreement with those reported previously using the X-ray film method (ThO<sub>2</sub> [3], MgO [4] and Y<sub>2</sub>O<sub>3</sub> [4]).

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## References

- 1. Y. WASEDA, K. HIRATA and M. OHTANI, *High Temp. High Press.* 7 (1975) 221.
- 2. B. D. CULLITY, "Elements of X-ray Diffraction" (Addison Wesley, Reading, Mass, 1956).
- H. HOCH and A. C. MOMIN, *High Temp. High* Press. 1 (1969) 401.
- 4. S. STECURA, Rev. Sci. Instrum. 39 (1968) 760.



Figure 1 Linear thermal expansion of  $ThO_2$ , MgO and  $Y_2O_3$ .

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